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# Francisco Jara,<sup>a</sup> Moisés Domínguez,<sup>a</sup> Marcos C. Rezende,<sup>a</sup> Edward R. T. Tiekink,<sup>b</sup>\* James L. Wardell<sup>c</sup> and Solange M. S. V. Wardell<sup>d</sup>‡

<sup>a</sup>Facultad de Quimica y Biologia, Universidad de Santiago, Casilla 40, Correo 33, Santiago, Chile, <sup>b</sup>Department of Chemistry, The University of Texas at San Antonio, 6900 North Loop 1604 West, San Antonio, Texas 78249-0698, USA, <sup>c</sup>Department of Chemistry, University of Aberdeen, Old Aberdeen AB24 3UE, Scotland, and, Instituto de Química, Universidade Federal do Rio de Janeiro, 21945-970 Rio de Janeiro, RJ, Brazil, and <sup>d</sup>Complexo Tecnológico de Medicamentos Farmanguinhos, Av. Comandante Guaranys 447, Jacarepaguá – Rio de Janeiro, RJ, Brazil

‡ Additional correspondence e-mail: solangewardell@yahoo.co.uk

Correspondence e-mail: edward.tiekink@utsa.edu

#### **Key indicators**

Single-crystal X-ray study T = 120 KMean  $\sigma$ (C–C) = 0.003 Å R factor = 0.055 wR factor = 0.142 Data-to-parameter ratio = 12.7

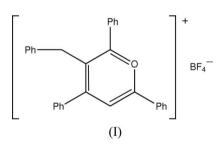
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved The central  $OC_5$  ring in the cation of the title compound,  $C_{30}H_{23}O^+ \cdot BF_4^-$ , has considerable aromatic character but the pendant aromatic rings are not coplanar. The crystal structure comprises undulating layers of cations separated by  $BF_4^-$  anions with significant intermolecular interactions between them.

3-Benzyl-2,4,6-triphenylpyrylium tetrafluoroborate

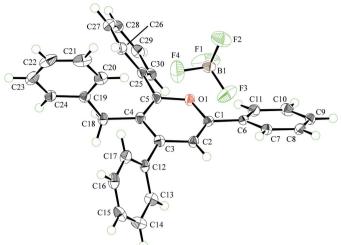
# Comment

The structure of the title compound, (I) (Fig. 1 and Table 1), shows significant twisting of the pendant aromatic rings out of the central plane. The dihedral angles between the O1/C1–C5 ring and the C6–C11, C12–C17 and C25–C30 rings are 28.14 (10), 56.70 (11) and 83.44 (10)°, respectively. Within the central ring, which formally carries a positive charge, the two O–C distances are equal within experimental error and the C–C distances lie in the relatively narrow range 1.361 (3)–1.414 (3) Å. These observations strongly suggest substantial delocalization of  $\pi$ -electron density over this ring.



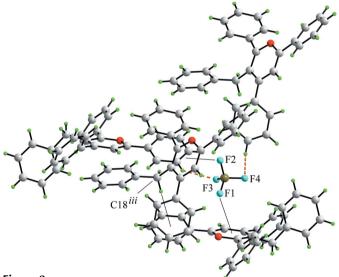
In the crystal structure there are a number of intermolecular interactions linking the ions. The primary interactions operating in the crystal structure are illustrated in Fig. 2. Here, B-F···H contacts are highlighted as golden dashed lines. The first contact occurs between the two components of the asymmetric unit so that  $C20-H20\cdots F4$  is 2.49 Å, C20···F4 is 3.355 (3) Å and the angle at H20 is  $152^{\circ}$ . The second  $F \cdot \cdot \cdot H$  contact involves the C8<sup>i</sup> and F3 atoms so that  $C8^{i} - H8^{i} \cdots F3$  is 2.55 Å,  $C8^{i} \cdots F3$  is 3.321 (3) Å and the angle at H8 is 139 Å [symmetry code: (i) 1 - x, 1 - y, -z]. The remaining two F atoms serve to link two central O1/C1-C5 rings. The parameters associated with these interactions are B1-F1···ring centroid(O1/C1-C5) = 3.091 (2) Å and angle at  $F1 = 114.52 (15)^{\circ}$ , and  $B1 - F2 \cdot \cdot \cdot ring centroid (O1/C1-C5)^{ii} =$ 3.080 (2) Å and angle at F2 =  $118.59 (17)^{\circ}$  [symmetry code: (ii)  $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$ ]. Formally, these might be considered as  $F \cdots$  ring centroid(O1/C1-C5) interactions, but as seen in the B-F...ring centroid(O1/C1-C5) angles there is significant bending in the angles so that F1 approaches atoms O1 and C1 at 2.9508 (24) and 2.9766 (28) Å, respectively, with the other  $F1 \cdot \cdot \cdot C$  distances being greater than 3.3 Å. A similar situation pertains for the interactions involving atom F2. Thus, the

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## Figure 1

The asymmetric unit of (I) showing 50% probability displacement ellipsoids and the atom-numbering scheme.



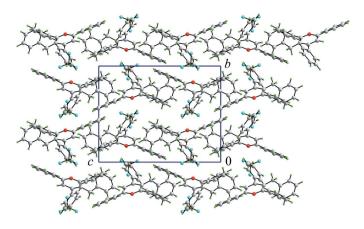
### Figure 2

Environment about the  $BF_4^-$  anion in (I) (Crystal Impact, 2006). Color code: O (red), C (grey), B (brown) and H (green).

F2...O1<sup>ii</sup>, F2...C1<sup>ii</sup> and F2...C5<sup>ii</sup> distances are 2.9270 (22), 3.1280 (28) and 3.1845 (26) Å, respectively, the remaining F2...C<sup>ii</sup> distances being greater than 3.5 Å. The only other significant intermolecular contact in the structure of (I) is also illustrated in Fig. 2, *i.e.* a C18<sup>iii</sup>—H18A<sup>iii</sup>…ring centroid(C6– C11) contact with an H18A<sup>iii</sup>…ring centroid distance of 2.98 Å and an angle of 152° at the H18A<sup>iii</sup> atom. The global crystal structure may be described as comprising undulating layers of cations interspersed with anions (see Fig. 3).

## **Experimental**

The title compound was isolated as a by-product in the preparation of 2-(4-hyroxyphenyl)-4,6-diphenylpyrylium tetrafluoroborate from 1,3diphenylpropenone and 4-hydroxyphenylethanone in the presence of boron trifluoride etherate (Aliaga *et al.*, 1997). The two products were separated by fractional crystallization from acetic acid. The title compound, recrystallized from AcOH, had melting point and spec-



### Figure 3

Packing diagram for (I), viewed approximately down the a axis (Crystal Impact, 2006). Color code as for Fig. 2.

troscopic properties in agreement with literature values (Marton et al., 1999).

Z = 4

 $D_x = 1.374 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Block, pale yellow

 $0.24 \times 0.10 \times 0.08 \text{ mm}$ 

25644 measured reflections 4136 independent reflections

3347 reflections with  $I > 2\sigma(I)$ 

 $\mu = 0.10 \text{ mm}^{-1}$ 

T = 120 (2) K

 $R_{\rm int}=0.042$ 

 $\theta_{\rm max} = 25.0^{\circ}$ 

#### Crystal data

 $C_{30}H_{23}O^+ \cdot BF^{4-}$   $M_r = 486.29$ Monoclinic,  $P2_1/c$  a = 10.6170 (3) Å b = 13.2065 (3) Å c = 17.1985 (5) Å  $\beta = 102.784$  (1)° V = 2351.68 (11) Å<sup>3</sup>

### Data collection

Bruker–Nonius KappaCCD diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 2003)  $T_{\min} = 0.834, T_{\max} = 1$ 

## Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0667P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	+ 2.1496 <i>P</i> ]
$wR(F^2) = 0.142$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} < 0.001$
4136 reflections	$\Delta \rho_{\rm max} = 0.80 \ {\rm e} \ {\rm \AA}^{-3}$
325 parameters	$\Delta \rho_{\rm min} = -0.40 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

## Table 1

Selected geometric parameters (Å, °).

01-C1	1.341 (3)	C2-C3	1.397 (3)
01-C5	1.350 (3)	C3-C4	1.414 (3)
C1-C2	1.361 (3)	C4-C5	1.381 (3)
C1-O1-C5	122.38 (17)		

All H atoms were allowed to ride on their parent atoms in the riding-model approximation at C–H distances of 0.95–0.99 Å, and with  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$ . Evidence of some disorder in the position of the BF<sub>4</sub><sup>-</sup> anion can be noted from Fig. 1. However, multiple positions could not be resolved.

Data collection: COLLECT (Hooft, 1998); cell refinement: DENZO (Otwinowski & Minor, 1997) and COLLECT; data reduc-

tion: *DENZO* and *COLLECT*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Crystal Impact, 2006); software used to prepare material for publication: *SHELXL97*.

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